

Characterization of LiNbO₃ Single Crystal Substrates Irradiated by Electron

電子線照射 LiNbO₃ 単結晶基板の評価

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Introduction

The lithium niobate (LiNbO₃) is a ferroelectric material. The LiNbO₃ crystal exhibits interesting piezoelectric, pyroelectric, electrooptic, and non-linear optical properties. Therefore, LiNbO₃ has various advantages for opto-electronic and acousto-optic applications including waveguides, modulators, second harmonic generators and surface acoustic wave (SAW) devices.¹⁾ However, large number of defects in LiNbO₃ are generated during crystal growth process, slicing process from ingots and wafer polishing. Thus, introduced defects are induced a large influence on several devices.²⁾

The electron irradiation will introduce the primary defect due to the atomic displacement damage in the material not influenced easily by the radiolysis. E. R. Hodgson and F. Agullo-Lopez reported that high-energy electron irradiation of stoichiometric LiNbO₃ induces F and F⁺ centers.³⁾ They describes that the oxygen vacancies also play an important role in LiNbO₃. However, there are few articles discussing at the polished surface and the non-polished surface.

In this study, we characterized the defects in electron irradiated LiNbO₃ single crystal substrates evaluated by photoacoustic spectroscopy (PAS), Fourier transform infrared (FT-IR) spectroscopy and Raman spectroscopy.

Experiments

The samples used were commercially available LiNbO₃ single crystal grown by conventional Czochralski method. The samples were cut out of 18 × 12 × 0.5 mm³ from a 128°-Y-cut substrate 50 mm diameter and one surface was mirror-polished. The electron irradiation with 1 MeV was carried out at JAEA, Takasaki. The samples were irradiated with electron at fluencies between 1 × 10¹⁶ and 1 × 10¹⁷ /cm² at a

fluence rate of 7.78 × 10¹² /cm²/sec. The electrons were irradiated onto the mirror-polished surface. To avoid sample heating up, samples were put on a water-cooled metal plate during electron irradiation. Thus, sample temperature was about 317 – 323 K during electron irradiation in this experiment.

The UV-VIS-IR absorption spectra of the samples were obtained with a gas-microphone photoacoustic spectroscopy (PAS) system,⁴⁾ operated in the 400-1600 nm range. Raman scattering spectra of samples were measured by a JASCO NRS-1000 micro laser-raman spectroscopy system. The wavelength of excitation laser is 532 nm using Nd:YVO₄. The infrared transmission spectra of the samples were obtained with a Fourier transform infrared (FT-IR) spectroscopy (JASCO FT/IR-610). All samples were measured at room temperature.

Results and Discussion

Photoacoustic (PA) spectrum of all samples are very broad. We normalized the spectrum based on the peak intensity of H₂O. Figure 1 shows the normalized photoacoustic (PA) signal intensity at 540 and 750 nm as a function of electron fluence. The peak of 540 nm and the peak of 750 nm show the electron color centers and O²⁻-hole centers, respectively.⁵⁾ The PA signal intensity at 540nm is not change by electron irradiation conditions. However, the PA signal intensity at 750 nm is increasing with increasing electron electron fluence. These results indicate that electron irradiation of our experimental range is affected for oxygen atoms.

IR transmission spectra around 3480 cm⁻¹ are shown in Fig. 2. The peak in this region indicates the OH⁻ absorption. The peak at around 3465 cm⁻¹ and the peak at around 3480 cm⁻¹ show the stoichiometric LiNbO₃ and Li-poor LiNbO₃, respectively.⁶⁾ Moreover, the OH⁻ absorption spectra are usually used to examine the LiNbO₃'s defect structure.⁷⁾ The shape of the spectrum is about the same compared with experiment condition at 5 × 10¹⁷ /cm². However, strength of the peak increases only in the experimental conditions at 5 × 10¹⁷ /cm². These

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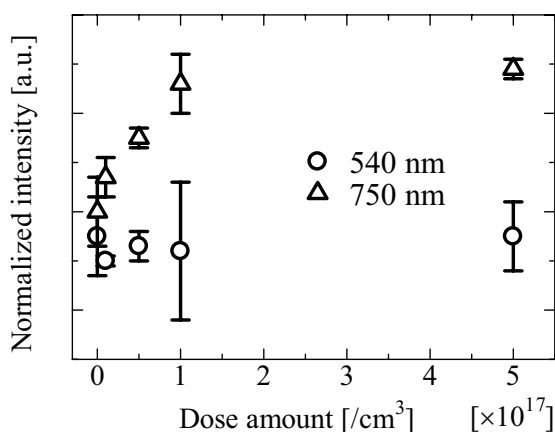


Fig. 1. PA signal intensity at 540 nm and 750 nm. The peak of 540 nm and the peak of 750 nm show the electron color centers and O^{2-} -hole centers, respectively.

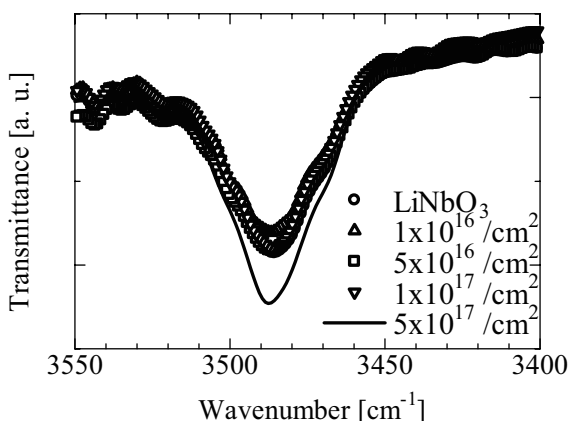


Fig. 2. FT-IR transmittance spectra related to the OH^- vibration. These spectra normalized by Nb-O vibration at around 1745 cm^{-1} .

results shows that defects in $LiNbO_3$ increase over $5 \times 10^{17}/\text{cm}^2$.

Raman spectroscopy is sensitive to the lattice straine and existance of point defects. Figure 3 shows the high energy (A_1 mode) Raman spectra of $LiNbO_3$ samples. There were no change in the Raman spectra on mirror-polished $LiNbO_3$ surface side (Fig. 3 (a)). However, all peak intensity are decreased with increasing electron irradiation conditions from non-polished $LiNbO_3$ surface (Fig. 3 (b)). We confirm that the mirror-polished side has less lithium (Li) than the non-polished side in X-ray photoelectron spectroscopy. Therefore, it is considered that the Li vacancy weakens the effect of the electron beam irradiation. Further work is underway to clarify electron irradiation effects for $LiNbO_3$, especially between surface composition and irradiation condisions.

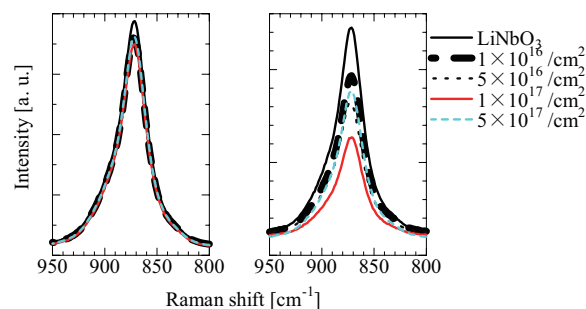


Fig. 3. Raman spectra of $LiNbO_3$ crystal irradiated onto mirror-polished surface. (a) The signal from mirror-polished surface, (b) non-polished surface.

Conclusions

We discuss about defects induced by electron irradiation in $LiNbO_3$. We understood that the electron beam irradiation introduced a defect, especially oxgen atom, in $LiNbO_3$ and mirror-polished surface, which has poor Li composition, is resists from electron irradiation effects.

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References

1. J. K. Yamamoto, T. Yamazaki, and K. Yamagishi, *Appl. Phys. Lett.* **64** (1994) 3228.
2. D. Xue and K. Kimura, *Solid State Commun.* **122** (2002) 537.
3. E. R. Hodgson and F. Agullo-Lopez, *J. Phys.: Condens. Matter* **1** (1989) 10015.
4. H. Takiguchi, Y. Okamoto, H. Miyazaki, and J. Morimoto, *Jpn. J. Appl. Phys.* **47** (2008) 3980.
5. I. W. Kim, S. S. Yi, V. F. Pichugin, V. Y. Yakovlev, and M. S. Demitriev, *J. Cryst. Growth* **253** (2003) 319.
6. H. -L. Wang, Y. Hang, J. Xu, L-H Zhang, S. -N. Zhu, and Y. -Y Zhu, *Mater. Lett.* **58** (2004) 31198
7. G. Bhagavannarayana, R. V. Ananthamurthy, G. C. Budakoti, B. Kumar and K. S. Bartwal, *J. Appl. Crystallogr.* **38** (2005) 768.